5(3) AUTHORS: Kochetkov, N. K., Rifantiyev, E. Ye., M. Story, L. V.

TITLE:

Bromination of B.-Ketoabetals (Bromirovaniya B-ketoatsetiley)

PERIODICAL:

Zhurnal obshchey kbimii, 1959, Vol 29, Mr 7, FF 2370-2337 (USSR)

ABSTRACT:

Hitherto the easily accossible β -ketoacetals RCOCH₂CH(OL'),

were used either as potential β -dicartenyl mapounds (note 1, 4), seeing they resemble the β -oblorovinylketones in their reactions or as ketones having another reactive substituent. A third possibility, i.e. that of introducing substituents into the central methylene group of the ketoacetal, has so far been disregarded, even though the resulting compounds could be utilized for synthesis in various directions (Ref 8). As first reaction of this kind the hitherto unknown bromination of the ketoacetals was undertaken. Thus, two methods were elaborated, one for the synthesis of α -brome- β -ketoaldehydes by bromination of β -ketoaldehydes in agreeus solution in the presence of barium carbonate, and another for the preparation of ethylene acetals of α -brome- β -ketoaldehydes in ether in the presence of barium carbonate. By contansation of α -brome- β -ketoaldehydes

Card 1/2

Bromination of β -Ketoacetuls

SOV/73-25-7-49/

with urea, the ?-amino-5-acyloxasoles were obtained; on condensing with thicures and thicamides of soils 2-substituted 5-acylthiazoles were formed. Thus, it was demonstrated, that the ∞ -bromo- β -ketoaldehydes react similarly to ∞ -bromoaldehydes in the reactions under investigation. On treating ethylene acetals of α -broms- β -kethaldehydes with β -naphthol in the presence of iron chloride and hydrochloric acid 2-alkyl-3-bromenaphtho-(1,2;5',6')-pyryl salts were formed. Fable 1 shows the synthesis of the x-bromo. B-ketoaldehydes:

R-CO-CHBr - C and table I the synthesis of the ethylene

acetals of x-brome- Aketoald-hydes:

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There are 2 tables and 15 references, 6 of which are Soviet.

ASSOCIATION: Moskovskiy gosudarstvennyy aniversitet (Moscow State University)

SUBMITTED: Card 2/2

June 5, 1958

5(3)

SOV/79-29-8-24/81

AUTHORS:

Kochetkov, N. K., Nifant'yev, E. Ye., Sokolov, S. D.

TITLE:

Synthesis of Aromatic β-Ketoacetals

PERIODICAL:

Zhurnal obshchey khimii, 1959, Vol 29, Nr 8, pp 2570-2575

(USSR)

ABSTRACT:

The β -ketoacetals which are mainly obtained from the corresponding β -chloro-vinyl ketones (Ref 1) are recently of manifold use in organic synthesis, whereas compounds of this class with aromatic radicals have nearly escaped notice (Refs 2-5). Quite recently, the authors synthesized the ethylene acetal of the benzoyl-acetaldehyde (Ref 6). Presently, they carried out the synthesis of various aryl- β -ketoacetals using, as initial products, the acyl- β -chloro-vinyl ketones which are now well accessible (Ref 7). Three new representatives of this class, the o-tolyl- and m-tolyl- β -chloro-vinyl ketone, and the n-bromo-phenyl- β -chloro-vinyl ketone, were synthesized. On the first attempts to obtain the acetal of benzoyl-acetaldehyde according to reference 1, the phenyl- β -methoxy-vinyl ketone (II) was formed instead of the expected compound. This was due to the catalytic action of alkali traces left over in the re-

Card 1/3

SOV/79-29-8-24/81

Synthesis of Aromatic β -Ketoacetals

action. Therefore, the reaction conditions were altered in such a way that in the distillation any traces of alkali were excluded by treating the reaction mixture with water, and extracting with ether. Thus the acetal of the benzoyl-acetaldehyde (III) resulted in a 60% yield. More convenient are the ethylene acetals of the β -ketoaldehydes which were recently synthesized by the authors (Ref 6). From among the representatives of the aromatic series, only the ethylene acetal of the benzoyl-acetaldehyde (IV) is described in publications. The fact that the cyclic ethylene acetals, which can easily be obtained by reaction of β -chloro-vinyl ketones with ethylene glycol, are formed by treating both the dimethyl acetals and the alkoxy-vinyl ketones with ethylene glycol in the presence of alkali, indicates their considerably higher stability. All transformations described, which are connected with the investigation of the stability of the acetals of the benzoylacetaldehyde, are illustrated in scheme 1. KOH and K2CO2 (Ref 6) proved to be the agents most useful for condensing the ethylene acetals of the β-ketoaldehydes of the aliphatic series. The operational method devised for the synthesis of the first member of the series was further applied to the syn-

Card 2/3

Synthesis of Aromatic β -Ketoacetals

SOV, 79-29-8-24, 61

thesis of the ethylene acetals of the β -ketoaldehydes substituted in the aromatic nucleus, using KOH and $K_2\text{CO}_3$ as con-

densing agents (60-80% yield):

Arco-cH= CHC1 \longrightarrow Arco-cH₂-c $\begin{vmatrix} c-cH_2 \\ 0-cH_2 \end{vmatrix}$

where $Ar = n, m, o-CH_3C_6H_5$; $n, o-ClC_6H_4$; $n-BrC_6H_4$; $n-CH_3o-C_6H_4$. The resultant crystalline acetals are stable, in general well soluble, and do not color with ferric chloride. There are 1 table and 14 references, 6 of which are Soviet.

ASSOCIATION:

Moskovskiy gosudarstvennyy universitet (Moscow State University)

SUBMITTED:

July 3, 1958

Card 3/3

5(3) AUTHORS:

Kochetkov N. K. Wifant yes E Ye. SOV/20-125-2-24/64

Kulakov, V N.

TITLE:

Synthesis of β Ketomercaptals (Sintez β -ketomerkaptaley)

PERIODICAL:

Doklady Akademii nauk SSSR 959, Vol 125, Nr 2, pp 327-329

(USSR)

ABSTRACT:

The preparative use of \$\beta\$ ketoacetals (Refs 1, 2) which can be obtained readily and with good yields from the interaction with alcohols and glycol of \$\beta\$ chlorovinylketones in an alkaline medium is rendered difficult by their very marked tendency towards hydrolysis in acid media. For this reason, the synthesis of the sulfurous analogues of the \$\beta\$ ketoacetals, i.e. of the substances mentioned in the title, was attempted. It was known that the mercaptal group is sufficiently stable in the acid medium (Ref 3). In view of the existing difficulties in the synthesis of pay methylene ketones (initial substances), the authors have developed a convenient general synthesis method for \$\beta\$ ketomeriaptals by means of ketovinylization of mercaptans (yields 50.90%). This reaction occurs quite readily in an aqueous solution in the presence of potash. As in the cases of the alcohols and of glycol (Refs 2), and unlike

Card 1/3

Synthesis of A-Ketomercaptals

SOV/20 125.2-24/64

the processes taking place in the cases of the phenols (Ref 5) and thisphenois (Ref 6) the reaction does not stop after the substitution of the chlorine atom in the chloroviny)ketone but is completed by the attachment of the second mercaptan morecule to the double bond. This is how mercaptal is formed. This reaction has a general character. On the one hand, this reaction is entered into by A ... chloroviny ketones both with aliphatic and with aromatic radicals on the other hand it is entered into by both monatomic and distomic met aptane. For this purpose, the sulfurous analogue of ethy: ene glycol. 1, 2-ethane dithiol (Ref 7) appears most appropriate. The aliphat: A-ketomercaptale thus produced are stable only liquids, their analogues with aremati. "adicals are solid well crystallizable substances. The ketomercaptals enter into such reactions as are typical of the Baketoaldehydes, which sufficiently proved the structure. They oxidize readily into the corresponding dieu fones (with perhydrol in HCl according to reference 81. These disulfones have a marked tendency towards hydrolythe decomposition in an alkaline medium. These reactions can be of interest for the production

Card 2/3

Synthesis of /3-Ketomercaptals

SOV/20-125-2-24/64

of various oxy-methyl-ketones. The experimental part contains the usual data. There are 2 tables and 13 references, 8 of

which are Soviet.

PRESENTED:

December 1, 1958, by A. N. Nesmeyanov, Academician

SUBMITTED:

November 29, 1958

Card 3/3

5.3730 5.3630

d1585 s/190/60/001, 1, B020/B066

AUTHORS:

Petrov, K. A., Nifant'yev, E. Ye., Fedorchuk, L. V.

TITLE:

Phosphorus-containing Polymers. I. Synthesis and

Polymerization of Ethylene Alkyl Phosphates

PERIODICAL:

Vysokomolekulyarnyye soyedineniya, 1960, Vol. 2. No. 3,

pp. 417-420

TEXT: Neutral phosphates are well-known compounds used in industry and agriculture. But so far no phosphates had been synthesized which contain in the molecule a five-membered cyclic ester grouping, nor high-molecular phosphates that had not been condensed via the vinyl ester groups. In the present paper, a simple process is suggested for the preparation of ethylene alkyl phosphates and their polymerization. These were synthesized by oxidation of ethylene alkyl phosphites with nitrogen dioxide. The cyclic phosphates synthesized were used by the authors to obtain polymers containing phosphorus. They also found that cyclic pnosphates (contrary to phosphonates) form higher-molecular compounds on

Card 1/2

NAME OF THE PROPERTY OF THE PR

Phosphorus-containing Polymers. I. Synthesis and Polymerization of Ethylene Alkyl Phosphates

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81585 \$/190/60/000 (07 /07 B020/B066

polymerization under the same conditions (with a degree of polymerization of 10 - 14). The resultant polymers had about the same molecular weight and the same properties as found in experiments without or with catalysts (sodium). High-molecular phosphates are neutral substances with polyester structure. The high-molecular phosphates described in this paper may be used as plasticizers. OThe experimental part describes the preparation of ethylene hexyl phosphite, ethylene phenyl phosphite, and ethylene alkyl phosphates (Table 1), the conditions of polymerization of ethylene alkyl phosphates and the properties of the polymers (Table 2) as well as the reaction of polyethylene propyl phosphate with phosphorus pentachloride. It is shown that ethylene alkyl phosphates polymerize on prolonged standing or heating. It was also found that the yield of ethylene alkyl phosphates obtained by the above-described process is 55 - 83 %, and that on polymerization the latter form polyesters (molecular weight of 2,000 - 3,000) by cleavage of the cyclic phosphates There are 2 tables and 5 references: 3 Soviet, 1 US, and 1 British.

SUBMITTED:

December 14, 1959

Cerd 2/2

83817

2109

S/190/60/002/005/007/C16 B004/B067

15.8114

Petrov, K. A., Nifant'yev, E. Ye., Sorikova, I. I.

AUTHORS:

Phosphorous Polymers. II. Use of the Arbazev Rearrangement

for Synthesizing Polyphosphonates

PERIODICAL:

Vysokomolekulyarnyye soyedineniya, 1960, Vol. 2, No. 1.

pp. 685-688

TEXT: The authors used the Arbuzov reaction for synthesizing polymer phosphinic esters. Polymerization occurs on heating 1 mole of cyclic phosphinites with 0.001 - 0.1 mole of methyl iodide in a sealed tube. Polyphosphinites are formed with a molecular weight of 270 - 3200. The hitherto unknown cyclic phosphinites were obtained by reacting dichlerophosphines with 1,3-diols in the presence of tertiary amines: phenylologically phosphinite and phenylology phosphinite. The reaction with CH₃I is a recurring alkylation reaction with

Card 1/2

83817

Phosphorous Polymers. II. Use of the Arbuzov Rearrangement for Synthesizing Polyphosphonates

S/190/60/002/005/007/015 B004/B067

$$\begin{array}{c} R & Q \\ CH_{3} & P-O \end{array} \left[-\left(CH_{2}\right)_{n} - P-O \right] \\ m - \left(CH_{2}\right)_{n} - I \text{ as final product. The structure of the} \end{array}$$

central member could be proved by cleaving the phenyl-(1.3-butylene) phosphinite polymer by means of phosphorus pentachloride. Phenyl (1 methyl-3-chloropropyl)phosphinyl chloride was obtained. The results of polymerization of phenyl-(1,3-butylene) phosphinite with various additions of methyl iodide are given in a table. There are 1 table and 2 references:

SUBMITTED:

January 15, 1960

Card 2/2

5.3600

177333 **S07/**73-30-1-57/75

AUTHORS:

Kochetkov, N. K., Nifant'yev, E. Ye., Nifant'yeva, L. V.

TITLE:

 $oldsymbol{eta}$ -Chlorovinyl Ketones of the Heteroey lie Series

PERIODICAL:

Zhurnal obshchey khimii, 1960, Vol 30, Nr 1, op del-

245 (USSR)

STATESTICAL PROPERTY OF THE PR

ABSTRACT:

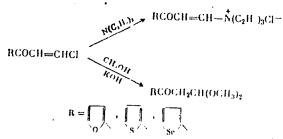
Synthesis of some β -chlorovinyl ketones, containing a five-membered heterocyclic radical, by the confensation of the corresponding acid chloriles with acetylene, was studied. It was found that acid chlorides of furance-carboxylic, thiophene-2-carboxylic, and selenophene-2-carboxylic acids easily condense with acetylene to form corresponding β -chlorovinyl ketones:

RCOCI + CH≡CH AICI RCOCII=CHCI

Card 1/4

Series Chlorovinyl Ketones of the Heterocyclic 773)3
Sov/79-30-1-2-773

The reaction takes place at $30\text{--}40^\circ$. The heterocyclic schlorovinyl ketones, like other vinyl ketones, react acetals:



They also readily condense with p-NO $_2$ C $_6$ H $_4$ NHNH $_2$ to form corresponding pyrazole derivatives. Thienyl-(a)-chlorovinyl ketone condenses with β -naphthol in the presence of ferric chloride and HCl.

Card 2/4

Series

Recollected Reconstruction Recollected Reconstruction Recollected Rec

Preparation of the following compounds is given: Fury1-(2)- β -chloroviny1 ketone (41%, based on action chloride), bp 102-105 (10 mm). Thieny1-(2)- β -chloroviny1 ketone (65%), bp 154-156.5 (23 mm). Seleny1-(2)- β -chloroviny1 ketone (45%), bp 132-135 (7 mm).

Card 3/4

G-Chlorovinyl Ketones of the Heterocyclic 77333 Series 77333

Dimethyl acetal of furoyl-(2)-acetaldehyde (0-%), be 122-123 (10 mm), n_D^{20} 1.4998, d20 1.1800. Dimethyl acetyl of thienoyl-(2)-acetaldehyde (5%), bp 1-7-148° (8 mm), n_D^{20} 1.5146, d_4^{20} 1.1910. 3-Furyl-(2')-1-(p-nitrophenyl)-pyrazole (62%), mp 70.5-72°. 5-Selenyl-(2')-1-(p-nitrophenyl)-pyrazole (63%), mp 100-101°. 2-Thienyl-(2')-napthy-(1,2:5,0)-pyrylium ferrichloride (66%), mp 176-177°. There are 11 Soviet references.

references

ASSOCIATION: Moscow State University (Moskovskiy gosudarstvennyy

universitet)

SUBMITTED: September 30, 1958

Card 4/4

S/079/60/030/006/032/033/XX B00⁻/B055

THE STATE OF THE S

AUTHORS:

Kechetkey N K and Mifant'yev, E. Ye

TITLE:

Oxidation of B-Ketoacetals by Means of Lead

Tetraacetate

PERIODICAL:

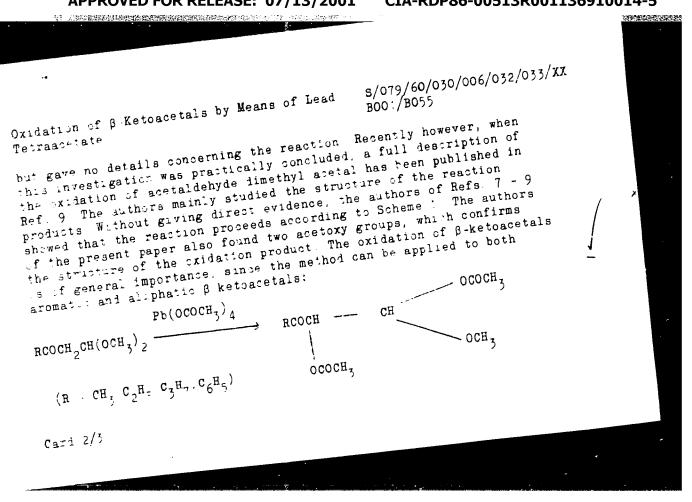
Zharnal obshchey khimii, '960 Vol 30 No. 6,

pp. 1866 - 1872

TEXT: The highly reactive and accessible $\beta\text{-ketoacetals}$ ROOCH_2CH(OR) _2 (Refs. - 4) are being used more and more in

synthetic chemistry though some of their very promising reactions have scarcely been investigated up to now. The reactivity of the central methylene group has been given least consideration, particularly as far as the substitution of its hydrogen atoms is concerned (Refs. 5.6). The present publication deals with the oxidation of β -keto-acetals by means of lead tetraacetate. Using this method, the authors (Ref. 3) and independently of them, other authors (Refs. 7.8) were able to introduce oxygen into the methylene group of β -ketoacetals,

Card 1/3



Hydrogenolysis of Tetrehydrofarans

s/079/60/030/006/033/033/xx BOO 1/BO55

are obtained which are not readily accessible by other methods, while or the second case only invaluable paraffin hydrocarbons are formed. There are 2 tables and 9 references: 4 Soviet 4 US and 1 British.

ASSOCIATION:

Institute organisheskiy khimii Akademii nauk SSSR (Institute of Organis Chemistry of the Academy of Sciences USSR)

SUPMITTED:

June 29 019

Carl 5/3

S/079/60/030/007/034/039/XX B001/B066

AUTHORS: Kochetkov, N. K., Nifant'yev, E. Ye., and Shibayev, V. N.

TITLE: Synthesis of Acyl-2-chloro-cyclohexenes-2 and Ethylene Ketals of 2-Acyl-cyclohexanones. A New Synthesis of Phen-

anthrenes (

PERIODICAL: Zhurnal obshchey khimii, 1960, Vol. 30, No. 7, pp. 2275-2282

TEXT: The authors describe the synthesis of the ethylene ketals of 2-acyl-cyclohexanones which have not been described as yet and were used as the starting material in a more convenient method of synthesizing phenanthrene derivatives. The synthesis was made on the basis of acyl-2-chlorocyclohexenes-2 which had been obtained by the authors in Ref. 1 by condensation of cyclohexanone with acid chlorides in the presence of AlCl₃, most

suitably in a molar ratio of 2-3 $AlCl_3$: 2-3 acid chloride: 1 ketone:

Card 1/4

Synthesis of Acyl-2-chloro-cyclohexenes-2 and Ethylene Ketals of 2-Acyl-cyclohexanones. A New Synthesis of Phenanthrenes

S/079/60/030/007/034/039/XX B001/B066

$$+ \text{RCOC1} \xrightarrow{\text{AlCl}_3} (R = CH_3, C_2H_5, iso-C_4H_9)$$

The reaction must be carried out at low temperature since otherwise resinification occurs (yield, 45-80%). On reaction of acyl-2-chlorocyclohexene-2/with ethylene glycol which has been earlier used by the authors (Refs. 2, 10, 11), the ethylene ketals of 2-acyl-cyclohexanones were obtained (50-60%)

Card 2/4

Synthesis of Acyl-2-chloro-cyclohexenes-2 S/079/60/030/007/034/039/XX and Ethylene Ketals of 2-Acyl-cyclo- B001/B066 hexanones. A New Synthesis of Phenanthrenes

The best solvent is dioxane. Ethylene ketals of 2-acyl-cyclohexanones in which one of the carbonyl groups is protected, are a convenient starting material. In this case, they were used as initial compounds for a new synthesis of the phenanthrene system. This synthesis is closely related to the synthesis of the naphthalene ring described by the authors in Refs. 10, 12, and is performed according to scheme 3. On reaction of the ethylene ketals with benzyl magnesium chloride, the corresponding oxy-ketals are formed which are directly converted to 1,2,3,4-tetrahydrophenanthrenes by aromatic cyclodehydration. The best condensing agents were hydrogen bromide in acetic acid, or mixtures of concentrated sulfuric and phosphoric acid. Tetrahydrophenanthrenes are separable by distillation. They are purified by producing the picrates. By this method, some 10-alkyl-1,2,3,4-tetrahydrophenanthrenes hitherto unknown were obtained in yields of between 25 and 55%. The structure of the resultant compounds was confirmed by the absorption spectra in ultraviolet, which are characteristic of the tetrahydrophenanthrene ring. The resultant tetrahydro-phenanthrenes are quantitatively converted to 9-alkyl-phenanthrenes when heated with palladium-on-carbon (Scheme 4). There are 19 references: 10 Soviet, 5 US, Card 3/4

Synthesis of Acyl-2-chloro-cyclohexenes-2 s/079/60/030/007/034/039/XX and Ethylene Ketals of 2-Acyl-cyclo- B001/B066 Synthesis of Phenanthrenes

1 British, 2 German, and 2 French.

ASSOCIATION: Moskovskiy gosudarstvennyy universitet

(Moscow State University)

SUBMITTED: July 6, 1959

Card 4/4

5.3400

5(3)

67951

AUTHORS:

SOV/20-130-1-26/69 Nifant'yev, E. Ye., Molodtsov, N. V., Kudryashov, L. I.,

Kochetkov, N. K.

THE REPORT OF THE PROPERTY OF

TITLE:

Ethylene Acetals of a Bromaroylacetaldehydes and Their

Transformations

PERIODICAL:

Doklady Akademii nauk SSSR, 1960, Vol 130, Nr 1, pp 94-97 (USSR)

ABSTRACT:

The authors wanted to synthesize eta-ketoazetals with functional groups in the molecule. For this purpose, they investigated the exchange reaction of the bromine atom in the α -bromo- β -ketoazetals RCO-CHBr-CH(GR)2 the synthesis method of which they had

worked out recently (Ref 2). \propto -Bromo-substituted ethylene acetals of the aromatic series $ArcochBrch(och_2)_2$ were best

suited. Such compounds were produced by bromination of the ethylene acetals of aroylacetaldehydes (see Scheme). The bromination was achieved either by bromine action in ethereal solution in the presence of barium carbonate (Ref 2) or by bromosuccinimide. The products obtained and mentioned in the title are stable, crystalline substances. Their bromine atom is quite readily exchanged by interaction with salts of some mineral acids. Thus, corresponding &-substituted ethylene acetals of

Card 1/3

57951

Ethylene Acetals of ∞ -Bromaroy those taldehydes and Their Transformations

SOV/20-130-1-26/69

aroylacetaldehydes (see Scheme) are formed, namely α -iodineand C-thiocyanogen-substituted ethylene acetals. A little more difficult is the substitution of browine by the nitro group while α -nitro- β -ketoacetal is formed. The above compounds represent a valuable initial material for the synthesis of some hardly accessible substances such as 4-benzoyl-2-oxythiazol. The interaction of brominated ketcacetals with mercaptanes proceeds smoothly. The reaction of the ethylene acetal of & -bromobenzoylacetaldehyde with sodiumbenzylmercaptide in methanol produces the ethylene aretal of α -benzylthiobenzoylacetaldehyde (see Scheme, Fig 1: 7 - the UV spectrum). The same bromoacetal reacts differently with sodium phenolate. No pure compound could be isolated from the resulting complex mixture by the reaction in acctone. On the other hand, the same reaction in methanol yielded a crystalline substance the analysis of which corresponded to the β -phenoxy- β -methoxy- α -oxy-hydrocinnamic aliehyde. Its UV spectrum (Fig 1: II) proves the missing benzoyl group and confirms the structure mentioned. It seems that the reaction with sodium phenolate proceeds via a transient &-oxide (similar to reactions described by T. I. Temnikova, Ref 5, see Scheme)

Card 2/3

Ethylene Acetals of ∞ -Bromaroylacetaldehydes and Their Transformations

67951 SOV/20-130-1-26/69

The interaction of bromoketoacetals with amines is complicated by the fact that - besides the exchange of the bromine atom - the acetal group enters the reaction. Thus, the phenyl- \propto , β -di-N-piperidylvinylketone develops in a high yield from the ethylene acetal of the \propto -bromobenzoylacetaldehyde and piperidine (UV spectrum, Fig 1: IV). Table 1 shows the constants and yields of the substances produced. There are 1 figure, 1 table, and 7 references, 5 of which are Soviet.

ASSOCIATION: Moskovskiy gosudarstvennyy universitet im. M. V. Lomonosova (Moscow State University imeni M. V. Lomonosov)

PRESENTED: June 9, 1959, by A. N. Nesmeyanov, Academician

SUBMITTED: June 6, 1959

Card 3/3

THE PROPERTY OF THE PROPERTY O

KOCHETKOV, H.K.; SOKOLOV, S.D.; VAGURTOVA, M.M.; HIFANT'YEV, E.Ye.

Organomagnesium compounds of the isoxazole series. Dokl. AN SSSR 133 no.3:598-601 Jl '60. (MIRA 13:7)

1. Moskovskiy gosudarstvennyy universitet imeni M.V. Lomonosova. Predstavleno akad. A.W. Mesmeyanovym.

(Magnesium organic compounds)

(Isoxazole)

KOCHETKOV, N.K.; NIFANT'YEV, E.Ye.

Chemistry of β -keto acetals. Usp. khim. 30 no. 1:31-47 Ja '61. (MIRA 14:2)

1. Khimicheskiy fakul'tet Moskovskogo gosudarstvennogo universiteta imeni M.V. Lomonosova.

(Acetals)

PETROV, K.A.; NIFART YEV, E.Ye.; NIKITINA, R.F.

Synthesis of diaryl phosphates and aryl phosphonates, and some of their properties. Zhur.ob.khim. 31 no.5:1705-1709 My '61.

(Phosphoric acid) (Phosphonic acid)

(MIRA 14:5)

PETROV, K.A.; NIFANT'YEV, E.Ye.; LYSENKO, T.N.

New synthesis of dialkyl phosphates. Zhur.ob.khim. 31 no.5:1709-1711 My '61.

(Phosphoric acid)

(Phosphoric acid)

PETROV, K.A.; NIFART'YEV, E.Ye.; GOL'TSOVA, R.G.

Interesterification of methyl phosphonites. Zhur.ob.khim.
31 no.7:2367-2370 Jl '61. (MIRA 14:7)

(Phosphonic acid) (Esterification)

PETROV, K.A.; NIFANT'YEV, E.Ye.; GOL'TSOVA, R.G.

Interesterification of monoethyl methylphosphinite with glycols.
Zhur.ob.khim. 31 no.?:2370-2373 Jl '61. (MIRA 14:7)
(Phosphinic acid) (Esterification) (Glycols)

PETROV, K.A.; NIFANT'YEV, E.Ye.; LYSENKO, T.N.; YEVDAKOV, V.P.

Synthesis of esters of phosphorous and phosphinic acids by alcoholysis of their amides. Zhur.ob.khim. 31 no.7:2377-2360 Jl '61. (MIRA 14:7) (Phosphorous acid) (Phosphinous acid)

"APPROVED FOR RELEASE: 07/13/2001

CIA-RDP86-00513R001136910014-5

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S/079/61/031/008/009/009 D215/D304

5.3630

Petrov, K.A., Nifant'ev E.Ye., Gol'tsova, R.G.

and Gubin, G.V.

TITLE:

AUTHORS:

Investigating the chemical properties of acid bisesters of ethylene glycol and methylphosphine acid

PERIODICAL:

Zhurnal obshchey khimii, v. 31, no. 8, 1961, 2732-2735

TEXT: In previous publications, the authors have shown that acid bismethylphosphonites could be prepared by esterification of the monomethylester of methylphosphinic acid with glycols. In the present investigations they studied some reactions of the simplest of these compounds, obtained by esterification with ethylene glycol. The following reactions were studied. 1) Oxidation of bismethylphosphinite [Abstractor's note: Called subsequently "the starting product"] with nitrogen oxides to the corresponding ester of bismethylphosphonic acid, according to scheme (NI). The obtained product is highly hygroscopic and reacts as a dibasic acid. 2) The reaction of the starting product

Card 1/2

25371 S/079/61/031/008/009/009 D215/D304

Investigating the chemical...

with sulfur; they did not succeed with the product itself, only with its sodium salt which was obtained from the product and sodium methoxide in dry methyl alcohol and could be isolated. (N2). 3) The reaction with dibutyldisulfide and methylthiocyanate (N3). 4) chlorination of the starting product which was successful with chlorine, but not with SO₂Cl₂; only a monochloride was obtained with chlorine which was oximized to a corresponding phosphonic acid (N4). 5) Aminomethylation with tetraethyldiaminomethylene; with equimolar amounts of reagents they obtained monomethyl diethylamino phosponate (N5). In the last two reactions the two phosphonic groups showed a different reactivity, only one of them taking part in the reaction. There are 5 references: 3 Soviet-bloc and 2 non-Soviet-bloc. The references to the English-language publications read as follows: L.W. Daasen, J.Am.Chem.Soc. 80,5301, 1958. E.K. Fields, J.Am. Chem.Soc. 74, 1528, 1952.

SUBMITTED: July 27, 1960

Card 2/2

S/079/61/031/009/001/012 D215/D306

AUTHORS: Petrov, K.A., Nifant'yev, E.Ye., and Khorkhoyanu, L.V.

TITLE: Peresterification of esters of dialkyl-phosphinious

acids with glycerine derivatives

PERIODICAL: Zhurnal obshchey khimii, v. 31, no. 9, 1961,

2889 - 2894

TEXT: In the present work the authors studied peresterification of dialkyl- and diarylphosphinious acids with glycerine derivatives containing one free hydroxyl group for use in insect repellant compounds. The reaction of 1,2-diphenylideneglycerine with 1,2-iso-propylideneglycerine was studied. The compounds were found to react readily with simpler esters, methylethyl-, dipropyl- and diphenylphosphinious acid. Glycerine derivatives with free secondary hydroxyls such as 1,3-benzylideneglycerine reacted less readily, but still gave good yields of the corresponding phosphinites. The phosphinites of the glycerine series provice novel compounds which

Card 1/6

Peresterification of esters ...

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S/079/61/031/009/001/012 D215/D306

are either liquids or crystalline solids having unpleasant odours. They oxidize easily in air but remain stable in an inert gas atmosphere; their chemical properties are similar to those of simpler dialkyl- and diarylphosphinious acids and in oxidizing medium and in the presence of sulphur convert to the corresponding phosphonates and thiophosphonates. The synthesized phosphinites react according to Arbuzov's reaction forming phosphine oxides and corresponding halogen derivatives. The peresterification and alkylation of phosphinites may be used in preparing some halogen derivatives from polyatomic alcohols if the former are difficult to produce. In the present work the authors also investigated this reaction in order to produce more complex halogen derivatives of the polyatomic alcohols. The propyl dipropylphosphinite and ethyl diphenylphosphinite necessary for this reaction were prepared by reacting Menshutkin acid chlorides with organomagnesium compounds at -700C

 $ROPCl_{2}' + R'MgB^{\bullet} \rightarrow ROPR_{2}'$ (1

Card 2/6

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Peresterification of esters ...

S/079/61/031/009/001/012 D215/D306

$$R = C_2H_5, C_3H_7,$$
 $R' = C_3H_7, C_6H_5.$ (1)

The first of the esters obtained has been unknown so far, and the second used to be prepared by more complex methods. The peresterification reaction was conducted by heating equimolecular quantities of the phosphinite and the glycerine derivative in a dry nitrogen stream with a small piece of sedium, distilling the required quantity of alcohol and finally vacuum distilling the residue. Time, temperature, yields and constants of the compounds obtained are given in tabulated form. In further experiments the propyl ester of dipropylphosphinious acid was oxidized with nitrogen oxides at -10°C until a permanent green coloration was obtained. Vacuum disphinic acid with high purity; b.pt. 103-104°C/1 mm Hg, nD - 1.4418, d4 - 0.9543, and having an empirical formula

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Peresterification of esters ...

S/079/61/031/009/001/012 D215/D306

 $^{\rm C}_9{^{\rm H}_{21}}{^{\rm O}_2}{^{\rm P}}$. The ester is colorless, odorless, insoluble in water, and soluble in ether, benzene, carbon tetrachloride and petroleum ether. Similarly oxidations of 1,2-isopropylidene glycerine ester and the 1,3-benzylideneglycerine ester of dipropylphosphinious acid were carried out to yield respectively esters of dipropylphosphinic acid, $^{\rm C}_{12}{^{\rm H}_{25}}{^{\rm O}_4}{^{\rm P}}$, b.pt. 143-144°/0.2 mm,

 $n_D^{20} - 1.4530$, $d_4^{20} - 1.0376$ and $C_{16}^H 2_1^0 _4^P$ b.pt. $117-118^0/10^{-4}$ mm $n_D^{20} - 1.5190$. Both esters are insoluble in water and petroleum ether and soluble in alcohol, acetone, benzene, chloroform and carbon tetrachloride. Addition of sulphur to both propyl- and 1,2-isopropylideneglycerine esters of dipropylphosphinious acid was conducted by heating the esters with thoroughly dry sulphur at $140-142^{\circ}C$ (exothermic reaction). The corresponding sulphur derivatives have b.pts. $81-82^{\circ}C/0.5$ mm and $141-140^{\circ}C/1$ mm respectively, unpleasant odors, and are both insoluble in water and soluble in

Card 4/6

S/079/61/031/009/001/012 D215/D306

Peresterification of esters ...

common organic solvents. Propyl-, 1,2-isopropylideneglycerine- and the 1,3-benzylideneglycerine esters of dipropylphosphinious acid undergo the Arbuzov rearrangement with methyl iodide to yield respectively dipropylmethylphospine oxide, b.pt. 91-93°C/l mm, m.pt. 39-39.5°C, the above oxide and 2,2-dimethyl-4-iodomethyldioxolen-1,3° b.pt. $81-83^{\circ}$ C/9 mm. n_{D}^{20} - 1.5038, and the oxide as before and

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2-phenyl-5-iodo-dioxan b.pt. $117-120^{\circ}\text{C/9}$ mm, n_D^{20} - 1.4983. The preparation of propyl ester of dipropylphosphinious acid and the ethyl ester of diphenylphosphinious acid was carried out by reacting the corresponding alkyl (propyl or ethyl) dichlorophosphite, pyridine, alcohol and propyl- or phenylmagnesium bromide respectively in ether at -65°C. Distillation of the reaction mixture yields in the first case the propyl ester of dipropylphosphinious acid b.pt. 70-71°C/7 mm, n_C^{20} - 1.4430, d_A^{20} - 0.8473 MR found 54.64; MR calculated 54.94, which is a mobile liquid with unpleasant smell which igni-

Card 5/6

Peresterification of esters ...

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\$/079/61/031/009/001/0₁₂ D215/D306

tes in air and which is insoluble in water but soluble in common organic solvents; in the second case the ethyl ester of diphenyl-phosphinious acid b.pt. 127-128°C/l mm, n_D²⁰ - 1.5910. There are 1 table and 8 references: 4 Soviet-bloc and 4 non-Soviet-bloc. The references to the English-language publications read as follows: U.S. Patent 2,588,407; E. Baer, H.L. Fischer, J. Am. Chem. Soc. 70, 609, 1948; C.N. Smith, D. Burnett, J. Econ. Entomol. 42, 434, 1949; T.H. Bevan, T. Malkin, D.B. Smith, J. Chem. Soc. 1955, 1383.

SUBMITTED: September 5, 1960

Card 6/6

PETROV, K.A.; NIFANT'YEV, E.Ye.; KHORKHOYANU, L.V.; TRUSHKOV, A.I.

Reesterification of esters of dialkyl- and diarylphosphinic acids.

Zhur.oh.khim. 31 no.9:3085-3090 S'61. (MIRA 14:9)

(Phosphinic acid) (Esterification)

PETROV, K.A.; NIFANT'YEV, E.Ye.; GOL'TSOVA, R.G.

Re-esterification of phosphinothioic and phosphonothioic esters with alcohols. Zhur.ob.khim. 31 no.10:3174-3177 0 '61.

(MTRA 14:10)

(Phosphonothioic acid) (Phosphinothibic acid) (Alcohols)

NIFANT'YEY, E YE. SOV/6034 PHASE I BOOK EXPLOITATION Konferentsiya po khimii i primeneniyu fosfororganicheskikh soyedineniy. 2d, Khimiya i primeneniye fosfororganicheskikh soyedineniy; trudy (Chemistry Kazan', 1959 and Use of Organophosphorus Compounds; Conference Transactions) Moscow, Izd-vo AN SSSR, 1962. 630 p. Errata slip inserted. 2800 copies printed. Sponsoring Agency: Akademiya nauk SSSR. Kazanskiy filial, Resp. Ed.: A. Ye. Arbuzov, Academician; Ed. of Publishing House: L. S. Povarov; Tech. Ed.: S. G. Tikhomirova. PURPOSE: This collection of conference transactions is intended for chemists, process engineers, physiologists, pharmacists, physicians, veterinarians, and agricultural scientists. COVERAGE: The transactions include the full texts of most of the scientific papers presented at the Second Conference on the Chemistry and Use of Card 1/14

43 SOV/6034 Chemistry and the Use of Organophosphorus (Cont.) Organophosphorus Compounds held at Kazan' from 2 Nov through 1 Dec 1959. The material is divided into three sections: Chemistry, containing 67 articles; Physiological Activity of Organophosphorus Compounds, containing 26 articles; and Plant Protection, containing 12 articles. The reports reflect the strong interest of Soviet scientists in the chemistry and application of organophospherus compounds. References accompany individual reports. Short summaries of some of the listed reports have been made and are given below. TABLE OF CONTENTS [Abridged]: Introduction (Academician A. Ye. Arbuzov) TRANSACTIONS OF THE CHEMISTRY SECTION Gefter, Ye. L. [NII plastmass (Scientific Research Institute of Plastics, Moscow]. Some Prospects for the Industrial Use of Organophosphorus 46 Compounds Card 2/14

Chemistry and the Use of Organophosphorus (Cont.)

SOV/6034

substituted phosphoric and phosphoric acids, as well as phosphorus-containing catalysts, have been synthesized and studied.

Petrov, K. A., V. A. Parshina, and G. L. Daruze. Phosphorus-Containing Polyester and Polyamide Resins
Bis-(p-carboxyphenyl)phosphonic acid, its esters, and salts, as well as amides and chlorides, have been obtained and for the first time described in the scientific literature. Organophosphorus polyesters and polyamides based on ethylene glycol, diethylene glycol, hexamethylenediamine, and bis-(p-carboxyphenyl)phosphonic acid

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Petrov, K. A., E. Ye. Nifant'yev, and I. I. Sopikova. Application of Arbuzov's Rearrangement to the Synthesis of Polyphosphonates

Previously unknown cyclic phenyl phosphonites have been obtained by the interaction of phenyldichlorophosphine with 1, 3-glycol. The phosphonites polymerize under the action of methyl iodide,

and capable of forming fibers have been obtained and described.

292

Card 7/14

S/190/62/004/002/012/02* B110/B101

AUTHORS:

Petrov, K. A., Nifant'yev, E. Ye.___

TITLE:

Phosphorylated polysaccharides. I. Phosphorylation of cellulose by transesterification of esters of acids of tra

valent phosphorus

PERIODICAL:

Vysokomolekulyarnyye soyedineniya, v. 4, no 2, 1962.

242 - 245

TEXT: Polysaccharides are phosphorylated by transesterification of esters of acids of trivalent phosphorus (neutral and acid phosphites and acid methyl phosphites). After 25-hr heating of diethyl phosphite together with anhydrous cellulose to $160-165^{\circ}\text{C}$ in the presence of metallic sodium, the P content in the end product reaches only 2.1%, and under harder conditions $175-185^{\circ}\text{C}$, 6.4%. The phosphorylated polysaccharide dissolves in excess diethyl phosphite to a gelatinous substance, and is separated again by water addition. The optimum temperature of reaction with triphenyl phosphite is 100°C (P content 5-6%). In the reaction of cellulose with monomethyl ester (I) or monoethyl ester (II) of methyl phosphinous acid at 100°C and Na catalyst, the P content in the end product is 19.1%. One, two, or Card 1/2

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Phosphorylated polysaccharides...

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three ester alkoxyls may react with the OH groups of cellulose. Since 2 05g of inulin are also phosphorylated by 8 g of II by 40 hr heating at 170 \pm 30 (23% P; V > 200), the method is of universal importance. The esters look like cellulose, are relatively inseparable (cellulose phosphinite with V = 150 does not burn), are soluble in trimethyl benzyl ammonium hydroneroxide and warm H_3PO_4 , partly swelling in organics, and partly hydrolyzing on prolonged treatment with H_2O_4 . Reactions reveal the presence of trivalent phosphorus. After prolonged at the state of the process of the process

phosphorus. After prolonged standing oxidation to pentavalent, hitherto unknown derivatives occurs. There are 10 references: Soviet and 9 non-read as follows: K. Weller, Canad Text. J. 70, 75 1953; R. F. Sinwenker, E. Pasck, Industr. and Engng. Chem., 50, 91, 1958; G. P. Toney US Patert 2759924, Chem. Abstr., 51, 713, 1957; S. Hobart et al., Text. Res. J. 21, 884, 1959.

SUBMITTED: February 9, 1961

Card 2/2

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133382 S/190/62/004/002/013/021 B110/B101

AUTHORS:

Petrov, K. A., Nifant'yev, E. Ye., Khorkhoyanu, L. V., Merkulova, M. I., Voblikov, V. F.

TITLE:

Phosphorus-containing polymers. III. Application of the Arbuzov reaction for polymerizing ethylene alkyl phosphites

PERIODICAL: Vysokomolekulyarnyye soyedineniya, v. 4, no. 2, 1962, 246-249

TEXT: The method by A. Ye. Arbuzov et al. (Izv. AN SSSR, Otd. khim. n., 1950, 357) can be used for producing polyphosphonates from cyclic phosphinites. In the present study, polyphosphonates were similarly synthesized on the basis of ethylene alkyl phosphites (I). Alcohol was acded dropwise to 126.5 g of ethylene chlorophosphite, 300 ml of ether, and 152 g of triethylamine; the mixture was left standing, filtered off, heated for 30 min, and (I) was obtained by double distillation. Cyclic phosphites contain an alkoxy group besides the cyclic ester group. Polyphosphonates are formed under catalytic action of methyl iodide on ethylene alkyl phosphite during 3 hr heating at 130°C in Ar atmosphere:

Card (1/3)

33;82 \$\frac{5}{190}/62/004/002/013/021 \$\text{B}110/\text{B}101

Phosphorus-containing polymers...

$$CH_{2}O \longrightarrow P - OR \xrightarrow{CH_{2}J} CH_{2} - P - OCH_{2}CH_{1}J \xrightarrow{OH_{2}O} POR$$

$$\rightarrow CH_{3} - P \longrightarrow OCH_{2}CH_{2} - P \longrightarrow OCH_{2}CH_{2}J$$

The structure of polyethylene heptyl phosphite was proven as follows:

$$\begin{bmatrix} -\operatorname{OCH_3CH_2} - \frac{1}{0} \\ 0 \end{bmatrix}_n \xrightarrow{\operatorname{PCl_4}} n \operatorname{ClCH_3CH_3P} - \operatorname{Cl} + n \operatorname{ClC_7H_{15}}$$

The degree of polymerization depends on the CH_3I amount, the reaction time and temperature. Optimum was: (1) small CH_3I amount; (2) $\sim 20-30$ hr, the reaction time depending on the molecular weight of the monomer, Card 2/3

Phosphorus-containing polymera...

33382 S/190/62/004/002/013/021 B110/B101

the reaction temperature, and the CH₃I concentration; (3) ~ 160 - 200°C, depending on the molecular weight (hexyl and isocctyl compounds: 160 - 170°C; nonyl and decyl compounds: 200°C). The polymers are viscous, colorless, and odorless liquids soluble in organics. Some of them are highly thermostable (polydecyl ethylene phosphite endures < 200°C for 20 - 30 hr). Utilization as plasticizer or admixture to lubricants is

$$\begin{array}{c|c}
CH_3 - O \\
CH_3 - O
\end{array}
\xrightarrow{PCl + HOR} \xrightarrow{N(C_2H_4)_4} CH_3 - O \\
CH_3 - O
\end{array}
\xrightarrow{PCH_3 - O} P - OR$$

was also synthesized. There are 2 tables and 5 references: 4 Soviet and 1 non-Soviet. The reference to the English-language publication reads as follows: A. K. Sherrill, J. Amer. Chem. Soc., 52, 1985, 1930.

SUBMITTED: February 9, 1961

Card 3/3

PETROV, K.A.; YEVDAKOV, V.P.; BILEVICH, K.A.; RADCHENKO, V.P.; NIFANT'YEV, E.Yo.

Properties of phospherus acid amides. Part 1: Reactions of amidophosphites with phenols. Zhur.ob.khim. 32 no.3:920-923 Mr '62. (MIRA 15:3) (Phosphoramidous acid) (Phenols)

PETROV, K.A.; NIFANT'YEV, E.Ye.; SHCHEGOLEV, A.A.

Glucose phosphinites. Zhur.ob.knim. 32 no.3:1006 Hr '62.

(MIRA 15:3)

(Glucose) (Phosphinic acid)

PETROV, K.A.; NIFANT'YEV, E.Ye.; GOL'TSOVA, R.G.; BELAVENTSEV, M.A.; KORNEYEV, S.M.

Esterification of phosphorous and phenylphosphinic acids. Zhur, ob.khim. 32 no.4:1277-1279 Ap '62. (MIRA 15'4)
(Phosphorous acid) (Phosphinic acid) (Esterification).

PETROV, K.A.; NIFANT'YEV, E.Ye.; SHCHEGOLEV, A.A.; KHUDYNTSEV, N.A.

Synthesis and chemical properties of phosphinites of 1,4;3,6—
dianhydrohexitol. Zhur.ob.khim. 32 no.9:3074-3080 S 162.

(Hexitol) (Phosphinic acid) (MIRA 15:9)

CIA-RDP86-00513R001136910014-5 "APPROVED FOR RELEASE: 07/13/2001

5/079/62/032/011/008/012 D204/D307 Petrov, K.A., Nifant'yev, E.Ye., and Gol'tsova, R.G. Peresterification of phosphites and phosphinites with 5 220 AUTHORS: Zhurnal obshchey khimil, v. 32, no. 11, 1962, substituted alcohols TEXT: The peresterification of the simpler esters of phosphorous TITLE: TEXT: The peresterification of the simpler esters of phosphorous acids was stuacid and methyl-, phenyl-, and dipropylphosphinous furfuryl and tedied, using amino-ethanol, halo- and cyanhydrins, trahvdrofurfurvl alcohols and with methyl cellosolve since the litrahvdrofurfurvl alcohols and with methyl cellosolve. PERIODICAL: died, using amino-ethanol, naio- and cyannydrins, luriuryl and territure concerning such reactions is very scarce. In a tunical externiture concerning such reactions is very scarce. In a tunical externiture concerning such reactions is very scarce. trangaruturrury alconors and with methyl cellosolve since the literature concerning such reactions is very scarce. In a typical exterature concerning such reactions is very scarce. In a typical exterature concerning such reactions is very scarce. periment a mixture of the ethyl ester of the phosphorous (or phosphinous) acid and the substituted sloobel was heated under an or other phinous) acid and the substituted sloobel was heated. periment a mixture of the ethyl ester of the phosphorous (or phosphorous) acid and the substituted alcohol was heated, under an inert atmosphere, to 150-1850C, until the calculated quantity of the calculated quantity. phinous) acid and the substituted alcohol was neated, under an officer atmosphere, to 150-1850C, until the calculated quantity of inert atmosphere, to 150-1850C, until the calculated quantity of inert atmosphere, to 150-1850C, until the calculated quantity of inert atmosphere, to 150-1850C, until the calculated at that temperative the tild distilled at the corresponding phosphile or phosphile of the subject to give the corresponding phosphile or phosphile. ture for a further 10-12 min, at 20-40 mm ng, and was then distilled to give the corresponding phosphite or phosphinite of the sub-led to give the corresponding phosphite or phosphinite. Furfury -diled to give the corresponding phosphite or phosphinite of the substituted alcohol. Na or H₂PO₄ were used as catalysts. Furfuryl-di-

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Card 1/2

S/079/62/032/011/008/012 D204/D307

Peresterification of phosphites ...

propyl- and - cyanoethyldipropylphosphinites were quantitatively oxidized to the corresponding phosphonates with a current of dry 0_2 . The rates and yields of the peresterifications were lower when the substituents in the alcohol were more electrophilic and when they were closer to the OH-carrying carbon atom. These data are in agreement with the mechanism showed earlier by the authors, i.e.:

(RO)₃P $\xrightarrow{\text{R'OH}}$ $\left[(\text{RO})_3 \text{P} \xleftarrow{\text{H}} \text{OR'} \right] \xrightarrow{-\text{R'OH}} (\text{RO})_3 \text{P} \xrightarrow{\text{ROH}} (\text{RO})_2 (\text{R'O}) \text{P}$

Those initial phosphites and phosphinites which possessed more strongly electrophilic substituents reacted more readily with the alcohols. Thus di- β -chloroethylphosphite and di- β -fluoroethylphosphite were reacted with decyl alcohol, at respectively 140-150°C and 120-130°C, in the presence of H₃PO₃, to give didecylphosphite in 80 and 85 % yields. There is 1 table.

SUBMITTED: December 14, 1961

Card 2/2

S/079/62/032/011/009/012 D204/D307

AUTHORS: Petrov, K.A., Nifant'yev, E.Ye., and Khorkhoyanu, L.V.

Phosphorylation of glycerine and its derivatives by alcoholysis of the amides of dialkylphosphinous acids. A new method of directed replacement of a hydroxyl by

a cyano group

TITLE:

PERIODICAL: Zhurnal obshchey khimii, v. 32, no. 11, 1962,

3720 - 3723

TEXT: Interactions of the diethylamide of dipropylphosphinous acid (I) with 1,2-<u>iso</u>-propylideneglycerine (II), 1,3-benzylideneglycerine (III) and glycerine were studied, in continuation of earlier work (ZhOKh, 31, 2889, 1961). I and II, and I and III interacted readily at 120-125°C to yield respectively the dipropylphosphinites of 1,2-<u>iso</u>-propylideneglycerine and 1,3-benzylideneglycerine (IV and V), in almost quantitative yields. Glycerine reacted analogously, at 135-140°C, in 60 % yield, to give the corresponding trisdipropylphosphinite (VI). C₃H₇OP (OC₃H₇)₂ reacted readily with bu-Card 1/2

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Phosphorylation of glycerine and ...

tyl thiocyanate, at 0°C, under an inert atmosphere; when the exothermic reaction was over the mixture was heated at 100-110°C for 1 hr., and was then distilled to give BuSP(0)(Pr)₂. Cyanodesoxy-1,2-<u>iso</u>-propylideneglycerine (VII) was prepared from IV and benzyl thiocyanate, in 45 % yield, by an analogous reaction. VII was converted into <u>iso</u>-propylidene-2-desoxytetrose by mixing it into SnCl₂/ether/HCl, stirring for 1 hr. at the b.p. of the ether evaporating the latter off, adding water and neutralizing the mixture and heating for 5 hrs. at 50°C. <u>Iso</u>-propylidene-3,4-desoxy-4-amino-erythrite was obtained by adding VII to ethereal LiAlH₄ evaporating the ether off, adding an equivalent amount of THF and heating for 96 hrs. on a water-bath.

SUBMITTED: December 14, 1961

Card 2/2

"APPROVED FOR RELEASE: 07/13/2001 CIA-RDP86-00513R001136910014-5 8/079/62/032/011/010/012 D204/D307 Petrov, K.A., Nifant'yev, E.Ye., Gol'tsova, R.G., Shchegolev, A.A., and Bushmin, B.V. Synthesis and peresterification of diphenyl phosphite 63630 Zhurnal obshchey khimii, v. 32, no. 11, 1962, 3723 - 3727 AUTHORS: TEXT: The interactions of diphenyl phosphite with aliphatic alco-TEAT: The interactions of diphenyl phosphite with allphatic simple hols were studied since the alcoholysis of diethyl and other in phosphites (to higher phosphites) and phosphites (to higher phosphites) TITLE: hols were studied since the alcoholysis of diethyl and Other simphosphites (to higher phosphites) and phosphinites requires, in some cases, inconveniently high temperatures (this journal, p. 1) phosphites (to higher phosphites) and phosphinites requires, in 3716, some cases, inconveniently high temperatures (this journal, p. C8 liso-C5H11, C6H13, C8 liso-C5H11, C6H13, Where R = C4Hg, iso-C5H11, C6H13, C8 liso-C5H11, C PERIODICAL: H₁₇, C₉H₁₉, C₁₀H₂₁, and C₂H₅OC(0)CH₂, were prepared in "17' '9"19' '10"21' ----2"2' moles ROH to 1 mole (PhO)2POH and heat91-96 % yields by adding 2 moles ROH to 1 mole ing for 3-8 hours at 100°C, in the presence or absence of catalyst ing for 3-8 hours at 100°C, in the presence or absence of catalyst to (Na). The high reactivity of diphenyl phosphite as compared to those of simple dialkyl phosphites is ascribed to (1) the existence card 1/2 Card 1/2

Synthesis and peresterification ...

S/079/62/032/011/010/012 D204/D307

of transistory forms $\begin{bmatrix} PhO \\ PhO \end{bmatrix}$ $P \leftarrow \begin{pmatrix} OR \\ H \\ OH \end{bmatrix}$ and $\begin{bmatrix} PhO \\ RO \end{pmatrix}$ $P \leftarrow \begin{pmatrix} OR \\ H \\ OH \end{bmatrix}$, which preference

rentially eliminate PhOH rather than ROH, owing to the considerably higher electrophilic character of the PhO group, and (2) the fact that the tautomeric equilibrium favors the trivalent P form far more in diphenyl than, say, in diethyl phosphite. Similar reactions took place readily with substituted alcohols such as e.g. (CH₃)₂

NCH₂CH₂OH. Diphenyl phosphite was obtained almost quantitatively by the equimolar interaction of diphenyl chlorophosphite with methanol (sealed tube, 100° C, 3 hrs.) and by the interaction of methyl dichlorophosphite with phenol (1:2) at 100° C for 1 hr. The latter method, which is generally convenient for the preparation of diaryl phosphites, was also used to make di-p- and di-m-cresyl phosphites, in ~ 100 % yields, by reacting CH₃OPCl₂ with para- and meta-cresols There is 1 table.

SUBMITTED:

December 14, 1961

Card 2/2

ACCESSION NR: AT4033987

\$/0000/63/000/000/0068/0072

AUTHOR: Petrov, K. A.; Nifant'yev, E. Ye.; Gol'tsova, R. G.; Kornayev, S. M.

TITLE: Polymers containing phosphorus. IX. Synthesis of acid polyalkylene phosphites, phosphates and thionphosphates

SOURCE: Geterotsepny*ye vy*sokomolekulyarny*ye soyedineniya (Heterochain macro-molecular compounds); sbornik statey. Moscow, Izd-vo "Nauka," 1963, 68-72

TOPIC TAGS: polymer, phosphorus containing polymer, polyalkylene phosphite, polyalkylene phosphate, polyalkylene thionphosphate, linear acid polyphosphite, polyphosphite synthesis, spatially discreet glycol, polyphosphite oxidation, polyphosphite alkylation

ABSTRACT: Linear acid polyphosphites were synthesized by reesterification of diethyl phosphite with spatially discreet glycols, then converted to polyalkylene phosphates by NO₂ oxidation or to thionphosphates by reaction with S. Successful syntheses (procedure described) were obtained with pentandiol-1,5, hexandiol-1,6, diethylene glycol, triethylene glycol, diethanolamine, pentafluoropentandiol-1,6, 1,4-3,6-dianhydrosorbitol, and p-dihydroxymethylbenzene. A neutral polythion-phosphite was obtained by alkylation of an ammonium salt of polyalkylenethionphosphoric acid. "We would like to thank S. A. Pavlova, associate at the INEOS AN SSSR

ASSOCIATION: AT4033987

for her help in determining the molecular weights." Orig. art. has: 2 graphs, 1 table and 3 chemical equations.

ASSOCIATION: none

SUBMITTED: 19Jun62 DATE ACQ: 30Apr64 ENCL: 00

SUB CODE: OC NO REF \$OV: 012 OTHER: 003

s/0000/63/000/000/0086/0089

ACCESSION NR: AT4017411

AUTHOR: Petrov, K. A.; Nifant'yev, E. Yee; Sopikova, I. I.; Merkulova, M. I.

TITLE: Chosphorylated polysaccharides. III. Phosphorylation of cellulose by dialkyl-(aryl)phosphites

SOURCE: Tsellyuloza i yeye proizvodnywye, sbornik statey (Cellulose and its derivatives). Moscow, 1963, 86-89

TOPIC TAGS: polysaccharide, cellulose, phosphorylated polysaccharide, cellulose phosphorylation, phosphorylation, dialkylphosphite, diarylphosphite

ABSTRACT: On the basis of the authors' previous work, the following studies were conducted: (1) phosphorylation of cellulose by di-β-chloroethylphosphite, di-βfluoroethylphosphite, and diphenylphosphite; (2) reaction of cellulose phosphite with tetraethylmethylenediamine; and (3) reactions of cellulose phosphite with chloral, diethyldisulfide, ethylsulfenechloride, and ethylthiocyanate. In the phosphorylation, 0.5 g of cellulose (cotton wool, thread and cord), dehydrated by washing with absolute alcohol, was reacted at 110, 130, 150 or 1650 for 30 or 60 hrs. with 25 g of the reagents in a stream of nitrogen at a pressure of 50-60 mm hrs. With 25 g of the reagents in a stream of nitrogen at a pressure of 50-60 mm hrs. With 25 g of the reagents in a stream of nitrogen at a pressure of 60-60 mm hrs. With 25 g of the reagents in a stream of nitrogen at a pressure of 50-60 mm hrs. With 25 g of the reagents in a stream of nitrogen at a pressure of 50-60 mm hrs. With 25 g of the reagents in a stream of nitrogen at a pressure of 50-60 mm hrs. With 25 g of the reagents in a stream of nitrogen at a pressure of 50-60 mm hrs. With 25 g of the reagents in a stream of nitrogen at a pressure of 50-60 mm hrs. With 25 g of the reagents in a stream of nitrogen at a pressure of 50-60 mm hrs. With 25 g of the reagents in a stream of nitrogen at a pressure of 50-60 mm hrs. With 25 g of the reagents in a stream of nitrogen at a pressure of 50-60 mm hrs. With 25 g of the reagents in a stream of nitrogen at a pressure of 50-60 mm hrs. With 25 g of the reagents in a stream of nitrogen at a pressure of 50-60 mm hrs. With 25 g of the reagents in a stream of nitrogen at a pressure of 50-60 mm hrs. With 25 g of the reagents in a stream of nitrogen at a pressure of 50-60 mm hrs. With 25 g of the reagents in a stream of nitrogen at a pressure of 50-60 mm hrs. With 25 g of the reagents in a stream of nitrogen at a pressure of 50-60 mm hrs. With 25 g of the reagents in a stream of nitrogen at a pressure of 50-60 mm hrs. With 25 g of the reagents in a stream of nitrogen at a pressure of 50-60 mm hrs. With 25 g of the reagents in a stream of nitrogen at a pressure of 50-60 mm hrs. With 25 g of the reagents in a stream of nitrogen at a pressure of 50-60 mm hrs. With 25 g of the reagents in a stream of nitrogen at a pressure of 50-60 mm hrs. With 25 g of the reagents in a stream of nitrogen at a pressure of 50-60 mm hrs. With 25 g of the reagents in a stream of nitrogen at a pressure of 50-60 mm hrs. With 25 g of the with methanol and ether and vacuum-dried over P_2O_5 . The AP and Cl content of 4.27-Cord 1/3

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ACCESSION NR: AT4017411	
8.56% and 2.97—9.06%, respectively, was found in the product, from cotton wool, while lower results were achieved with visco and cord. Di- β -chloro- and di- β -fluoroethylphosphites were fo be better suited for the reaction. In the reaction with tetra ethylmethylenediamine and disulfides, conversion of cellulose	ose thread ound to phos-
phites into a-hydroxy- and a-aminophosphonates and thiolophosp was also accomplished by the reactions	hates
cellulose OP OCH ₂ CH ₂ Cl (C ₂ H ₂) ₂ NCH ₂ N(C ₂ H ₄) ₃ cellulose OP OCH ₂ CH ₂ Cl CH ₂ Cl (C ₃ H ₃) ₄	r
cellulose OP OCH, CH, CI(C, H, SCN) OH Cellulose OP OCH, CH, CI SC, H,	II
Orig. art. has: 1 table.	
ASSOCIATION: none	-
Card 2/3	

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ACCESSION NR: AT4017411		:	
SUBMITTED: 12Apr62	ATD PRESS:	3045	ENCL: 00
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ACCESSION NR: AT4017412

s/0000/63/000/000/0090/0093

AUTHOR: Petrov, K. A.; Nifant'yev, E. Ye.; Sopikova, I. I.; Belavintsev, M. A.

TITLE: Phosphorylated polysaccharides. IV. A method for phosphorylating cellulose using phosphorous acid

SOURCE: Tsellyuloza i yeye proizvodny*ye, sbornik statey (Cellulose and its derivatives). Moscow, 1963, 90-93

TOPIC TAGS: polysaccharide, polysaccharide phosphorylation, cellulose, cellulose phosphate, phosphocellulose, phosphorylation

ABSTRACT: Cellulose was phosphorylated by phosphorous acid using 3 different procedures: (1) reacting cellulose and molten phosphorous acid at 100C for 10 hours in a current of dry nitrogen, yielding a product containing 15-17%P;(2) in dimethylformamide or o-xylene solutions in a series of 20 to 60-hour tests at 130 and 160C yielding a product containing 4.8-12.2%P; (3) prolonged (2-3 days) heating at 80-140C in an atmosphere of an inert gas, which proved to be the most suitable since it yielded products containing up to 14%P. Different kinds of cellulose were tested, and the one swollen in water or pyridine was found best. The P-content in the product increased with the concentration of phosphorous acid up to a certain limit, the optimal ratio being one in which there is slightly more than Card 1/2

ACCESSION NR: AT4017412

one phosphorous acid molecule for each /3 ,d-glucose unit in the reacting mixture. Orig. art. has: 2 graphs and 1 table.

ASSOCIATION: none

SURMITTED: 12Jul62

DATE ACQ: 06Jan64

ENCL: 00

SUB CODE: CH

NO REP SOV: 006

OTHER: 003

\$/0000/63/000/000/0170/0174

ACCESSION NR: AT4034002

AUTHOR: Patrov, K. A.; Nifant'yev, E. Ya.; Gol'tsova, R. G.

TITLE: Phosphorus-containing polymers. X. Synthesis of polyphosphite-based polyphosphonates

SOURCE: Geterotsepny*ye vy*sokomolekulyarny*ye soyedineniya (Heterochain macromolecular compounds); sbornik statey. Moscow, Izd-vo "Nauka," 1963, 170-174

TOPIC TAGS: polymerization, phosphorus containing polymer, polyphosphite, polyphosphonate

ABSTRACT: As a further step in the authors' polymer studies data are given on the synthesis of polyalkylalkylenephosphonates, poly-\$\mathcal{L}\$ -hydroxyalkylalkylenephosphonates by the Michaelis phosphonates and poly-\$\times\$-aminoalkylalkylenephosphonates by the Michaelis and Becker method using acid polyalkylenephosphites. The following polymers were prepared, identified and described: polybenzylhexamethylenephosphonate, poly-\$\times\$-dibutylaminobenzylhexamethylenephosphonate, poly-\$\times\$-dibutylaminobenzylhexamethylenephosphonate, poly-\$\times\$-propylenephosphonate, poly-\$\times\$-dibutylaminomethylhexamethylenephosphonate, poly-\$\times\$-encoded -propylenephosphonate, poly-diethylaminomethyl-p-xylidenephosphonate, polybutylamino-bis-methylhexamethylenephosphonate, and a copolymer of Card 1/2

ACCESSION NR: AT4034002

}:

d-dibutylaminobenzylhexamethylenephosphonate and di-(hexamethylenephosphato) disulfide. The preparative procedure consists essentially of reacting the reagents for several hours at 90-135C; the yield varied from 48 to 98% for different individual products. Orig. art. has: 4 chemical equations.

ASSOCIATION: None

SUBMITTED: 13Nov62

DATE ACQ: 30Apr64

ENCL: 00

SUB CODE: OC

NO REF SOV: 009

OTHER: 004

Card 2/2

ACCESSION NR: AT4034009

8/0000/63/000/000/0240/0242

Ċ.

AUTHOR: Petrov, K. A.; Mifant'yev, E. Ye.; Ly*senko, T. N.

SOURCE: Geterotsepny*ye yy*sokomolekulyarny*ye soyedineniya (Heterochain macromolecular compounds); sbornik statey. Moscow, Izd-vo "Nauka," 1963, 240-242

TOPIC TAGS: polymerization, polymer, phosphorus containing polymer, alpha propylglucoside, phenylglucoside, polyphosphinite

ABSTRACT: In an extension of the authors' previous work on phosphorus-containing polymers, a number of polyphosphites and polyphosphinites were obtained by the alcoholysis of phosphoamides and reesterification of arylphosphites and arylphosphinites, using \prec -propylglucoside and N-phenylglucoside as the reagents. In the alcoholysis procedure, 1 nol of \prec -mathyl, \prec -propyl or N-phenylglucoside and 2 or 2.5 mols of phosphoamide $(C_3H_7OP [N(C_2H_5)_2]_2$, $C_4H_9OP [N(C_2H_5)_2]_2$, $C_8H_17OP [N(C_2H_5)_2]_2$) were heated at 140-145C for 3 hrs., at 140-150C/10 mm for 4 hrs. and at 180-190C/3 mm for 3 hrs. in a stream of inert Card 1/2

ACCESSION NR: AT4034009

gas. Oxidation of the polyphosphites and polyphosphinites to polyphosphates and polyphosphonates with nitrogen dioxide was also conducted and the reaction of acid poly-N-phenylglucophosphite with chloral demonstrated. The polyglycophosphites and polyglycophosphinites obtained contain hydrophobic radicals and less thermo- and hydrolytically stable than the corresponding polyglycophosphates and polyglycophosphonates. Orig. art. has: 1 figure and 1 table.

ASSOCIATION: None

SUBMITTED: 24Apr63

DATE ACQ: 30Apr64

ENCL: 00

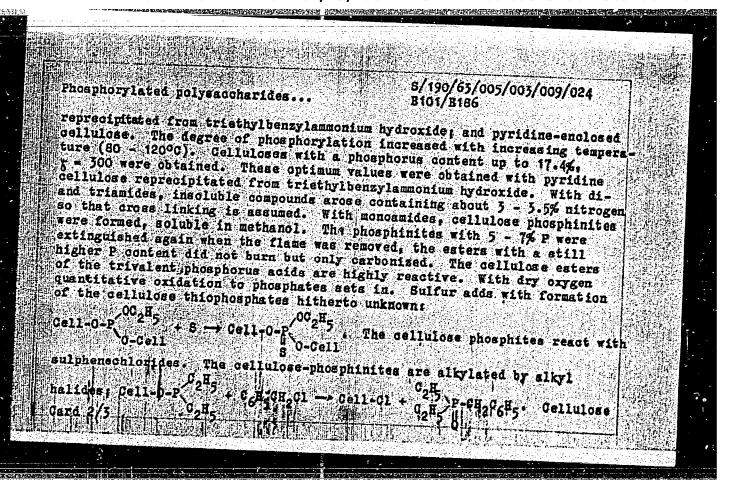
SUB CODE: OC

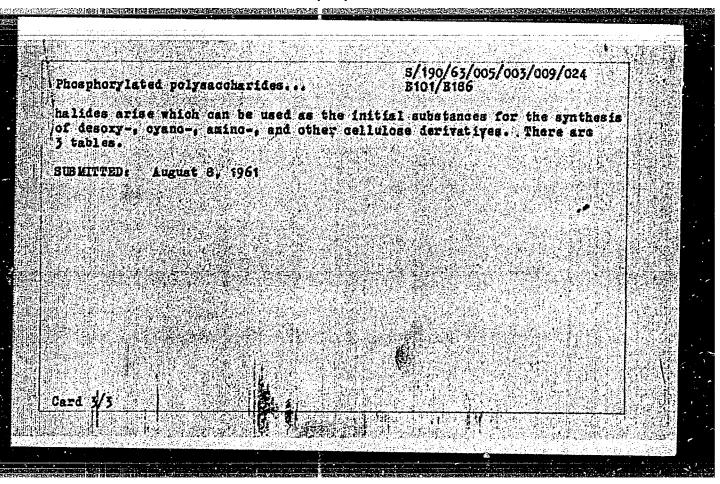
NO REF SOV: 005

OTHER: 003

Card 2/2

8/190/63/005/003/009/024 B101/B186 AUT HORS Petrov, K. A., Effant lyer, R. Ye., Khorkhoyanu, L. V., Yoblikay, V. F. TITLE Phosphorylated polysadcharides. II. Phosphorylation of dellulose by alcoholysis of amides of the acids of threevalent phosphorus PERIODICAL: Vysokomolekulyarnyje soyedinenija, v. 5. no. 3. 1963. 348-352 TEXT: In a previous paper (Zh. obshch. khimii, 51, 2377, 1961) the authors described the reaction: Cell-OH + R_2M - P< + Cell-O-P< + R_2MH . In the present paper a study was made of this new method of phosphorylating cellulese by elcoholysis of phosphorous soid smides such a diethyl phosphorus acid diethylamide, sthylphosphorous acid tetraethylemide and phosphorous acid hexaethyltrismide in order to develop fireproof, antiseptic and insecticidal cellulose. Since the process of esterification of cellulose depends to a great extent on how the sample is prepared the following cellulose types were subjected to phosphorylations viscose fiber, washed with methanol and dried; specially prepared cotton cellulose; cellulose Card 1/3





L 13552-63 EMP() ACCESSION NR: AP3000699 EMP(1)/EPF(c)/EMT(m)/BDS ASD PC-LIPE-L RM/WW 8/0190/63/005/005/0712/0718 AUTHOR: Petrov, K. A.; Mifant yev, R. Ye.; Ly senko, T. R.; Susanskiy, TPTE: Phosphorus-containing polymers. Synthesis of polyphosphites and polyphosphinites on the basis of glucose SOURCE: Vyfsokomolekulyarnyfys soyedineniys, v. 5, no. 5, 1965, 712-718 TOPIC TACS; polyphosphites, polyphosphinites, methylglucoside, phosphorylation, polymers, monosaccharides, polythiophosphates ABSTRACT: The objective of the present investigation was an attempt to synthesize polymers, using methylglucoside from the hydrolysis of wood pulp and di- and triamides of trivelent phosphorous scids as the issuing materials. However, the elcoholysis by methylglucoside of tetraethyldiamides of methylphosphinic and ethylphosphorous acid and of hexaethyltriamide of phosphorous acid, when conducted at 100 to 130C, yielded bicyclic phosphinites with a molecular weight of only 210 to 220. Subsequent heating at 160 to 2000 caused a molecular weight increase, with optimal molecular weights reaching 321.000 and 528.000, where the ratios of the reacting ingredients are close to equinormal. Since the formation of intermolecular bonds generally proceeds at a lesser rate when compared with the building of intracyclic phosphinite groups, it is necessary to conduct the polymerization in two steps, the Card 1/2

	is polythiophosphator, and polyph is at 1500, and with Arbuzov's all has: 2 formulas, 5 figures, and (ained polymers could be caphonates by treatment kylation reagent, 2 tables.
ASSOCIATION: none		
SUBSTITUED: 0180y61	DATE ACQ: 17Jun63	ERCL: 00
SUB CODS; CE	. No rest sov: 008	OTHER: 004

PETROV, K.A.; NIFANT!YEV, E.Ye.; GOTTSOVA, R.G.; SOLNTSEVA, L.M.

Phosphorus-containing polymers. Part 7: Synthesis of polyphosphites and polyphosphinites by glycolysis of amides of trivalent phosphorus acids. Vysokom.soed. 5 no.11:1691-1695 N '63. (MIRA 17:1)

PETROV, K.A.; NIFANT'YEV, E.Ye.; KHORKHOYANU, L.V.; GOL'TSOVA R.G.

Phosphorus-containing polymers. Part 8: Synthesis and some properties of polyarylene phosphites and phosphinites.

Vysokom. soed. 5 no.12:1799-1804 D '63. (MIRA 17:1)

PETROV, K.A.; NIFANT'YEV, E.Ye.; SHCHEGOLEV, A.A.

Synthesis of 1,2-dialkyl phosphinites; 5-6-diisopropylidensglucoses and their conversion to 6-halodeoxyglucose. Zhur.ob.khim.

33 no.3:896-899 Mr '63.

(Phosphinic acid)

(Glucose)

PETROV, K.A.; NIFANT'YEV, E.Ye.; SHCHEGOLEV, A.A.; BUTILOV, M.M.; REBUS, I.F.

Re-esterification of neutral phosphites and phosphinites.
Zhur.ob.khim. 33 no.3:899-901 Mr '63. (MIRA 16:3)
(Phosphinic acid) (Phosphorbus acid)
(Esterification)

PETROV, K.A.; NIFANT'YEV, E.Ye.; GOL'TSOVA, R.G.

Trans-esterification of diethyl phosphite with ethylene glycol.
Zhur. ob. khim. 33 no.5:1485-1488 My '63. (MIRA 16:6)

(\$thyl phosphites) (Esterification)
(Ethylene glycol)

NIFANT'IEW, E.Ye.; GRACHEV, M.A.; BAKINOVSKIY, L.V.; KARA-MURZA, S.G.;

KOCHETKOV, N.K.

Synthesis of methyl — -chlorovinyl ketone. Zhur.prikl.khim. 36 no.3:
676-678 My '63. (MIHA 16:5)
(Ketone) (Vinyl compounds)

PETROV, K.A.; MIFANT'YEV, E.Ye.; LIBMAN, B.Ya.

Synthesis of di-(2-ethylhexyl) phosphate and phenyldi(2-ethylhexyl) phosphate. Zhur. prikl. khim. 36 no.8:18531857 Ag 83.

(MIRA 16:11)

L 17550-63 EWP(j)/EPF(c)/EWT(m)/EDS Pc-L/Pr-L RM/WW

ACCESSION NRs AP3004425

8/0020/63/151/004/0859/0861

AUTHORS: Petrov, K. A.; Mifant yev, E. Ye.; Sopikova, I. I.

TITIEs Phosphorylation with acylphosphites.

SOURCE: AN SSSR. Doklady*, v. 151, no. 4, 1967, 859-861

TOPIC TAGS: phosphorylation, acylphosphite, alcohol, acid phosphonate

ABSTRACT: The purpose of this work was to develop new methods for phosphorylating hydroxyl compounds. Tribenzoylphosphite and butylene-1,5-acetylphosphite were used at relatively low temperatures in the presence of triethylamine. Tertiary phosphites were formed in high yields with primary, secondary, and tertiary alcohols and with acid phosphonates such as the monopropyl ester of methylphosphinic acid. They can also be used for phosphorylating carbohydrates and other natural products. The reaction of these acylphosphites with phosphoxanthogenates produced thiophosphates and other sulfur-containing compounds. These phosphorylating agents can be obtained readily by reacting PCI; or butylene-1,5-chlorophosphite with metallic salts of the corresponding carboxylic acid. Phosphoxanthogenates were produced by reacting PCI; and chlorophosphines with salts of alkylxanthogenic acids. Syntheses of the following are described: tributylphosphite; 1,5-butylene-tert.butylphosphite; 1,5-butylene-tert.butylphosphite; Card 1/2

ACCESSION NR: AP3004425 and 0-0-1-3-butylens-0-m	onelmathological	\mathcal{O} :
	copylmathylaubphosphonate. The plescribed. The original artic	reparation of tri-Alpha- Le has 3 formulas.
ASSOCIATION: none		
SUBMITED: 26Jan63	DATE ACQ: 21Aug63	ERCL: 00
SUB CODE: CH	NO REF SOVE COR	OTHER: 004
	Company of the Compan	
Gard 2/2		

POPOV, G.N.; NIFONTOV, B.I.; LOBANOV, D.P.; KULIKOV, A.V.;
KALYUZHNAYA, T.P., red.

[Characteristics of the development of radioactive ore deposits] Osobennosti razrabotki mestorozhdenii radioaktivnykh rud. Moskva, Atomizdat, 1964. 218 p.

(MINA 17:6)

1 34919-65

ACCESSION NR: AT5006104

8/0000/64/000/000/0042/0061

AUTHOR: Yerokhin, R. A.; Koshurnikova, N. A.; Lyubchanskiy, E. R.; Nifatov, A. P. Reshetov, G. N.

TITLE: Content and microdistribution of plutonium-239 in rat lung and liver and morphological changes in these organs after intratracheal administration of the isotope

SOURCE: Raspredeleniye, biologicheskoye deystviye, uskoreniye vyvedeniya radioaktivnykh izotopov (Distribution, biological effect, acceleration of the excretion of radioactive isotopes); sbornik rabot. Moscow, Izd-vo Meditsina, 1964, 42-61

TOPIC TAGS: plutonium-239, radioisotope, inhalation, liver, lung, pathology, radicactivity, lymphatic system

ABSTRACT: The behavior of plutonium in the lung following intratracheal administration of various salts is determined largely by the physicochemical form of the compound used. The plutonium content of the lungs after administration of the nitrate was 5-10 times higher than after administration of sodium plutonyl triacetate. The clearance of plutonium administered in the form of these two salts obeys the exponential law, but it was more rapid in the case of the second salt. A large quan-

Card 1/3

L 34919-65

ACCESSION NR: AT5006104

tity of plutonium was transported from the lungs by macrophages into the regional lymph nodes. Plutonium accumulated in the liver during the early phase (20 minutes to 24 hours) more slowly after administration of the nitrate than it did after administration of sodium plutonyl triacetate. During the later phases (4 to 6 months) the rate of deposition in the liver was about the same after administration of either form of plutonium - 0.90-0.55 and 0.95-0.57% of the dose administered.

The microdistribution of plutonium in rat liver after intratracheal administration of the two plutonium salts was quite diffuse. Histological changes in the lung varied with the nature of the microdistribution of the element and they arose mainly in the places where the isotope concentrated. The severity of the pathological changes and the time when they developed were related to the ionization dose that accumulated. Among the earliest changes were degeneration; desquamation of bronchial and alveolar epithelium, and perivascular edema. These were followed by chronic inflammation, chiefly productive in character. The pathological process developed into pneumosclerosis as a result of the proliferation of connective-tissue cellular elements with the formation of fibrous structures. No significant morphological changes were noted in the liver after intratracheal administration of 7 µc/kg of plutonium nitrate or sodium plutonyl triacetate. Orig. art. has: 15 figures, 2 tables.

Card 2/3

"APPROVED FOR RELEASE: 07/13/2001 CIA-RDP86-00513R001136910014-5

L 3h919-65 ACCESSION NR: AT50	06104		0	
ASSOCIATION: none				IS
SUBMITTED: 10Apr64		ENCL: 00	SUB CODE:	
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1_34113-55 EWG(1)/EWT(h) 05

ACCESSION NR: AT5006122

8/0000/64/000/000/0183/0191

AUTHOR: Nifatov, A. P.; Buldakov, I. A.

TITLE: Biological effects of strontium-96, cesium-137, and promethium-147 after chronic oral administration of low doses of the isotopes

SOURCE: Raspredeleniye, biologicheskoye deystviye, uskoreniye vyvedeniya radioektivnykh izotopov (Distribution, biological effect, acceleration of the excretion of radioactive isotopes); sbornik rabot. Moscow, Izd-vo Meditsina, 1964, 183-191

TOPIC TAGS: strontium-90, cesium-137, promethium-147, radioisotope, radioactivity, blood, viscera

ABSTRACT: Oral administration of 0.00051 and 0.012 $\mu c/24$ hours of Sr^{90} , 0.00124 $\mu c/24$ hours of Cs^{137} , and 0.0027 $\mu c/24$ hours of Pm^{147} to rats for almost two years had no significant effect on the average survival time, weight, peripheral blood, or rate of tumor formation in the rats. Although the overall number of tumors was the same in both the experimental animals and the controls, the spectrum of neoplasms was different. The experimental animals had more tumors of the adrenal cortex, testes, gastrointestinal tract, and generalized leukemias, a phenomenon not explained by the authors. Since the greater frequency of glandular tumors was not

Card 1/2

L 34113-65

ACCESSION NR: AT5006122

related to the dose or to local irradiation, the authors doubt that these neoplasms were the result of direct radiation injury or that they were the determining factor in the death of the animals. The main causes of death seemed to be various inflammatory processes in the lungs and gastrointestinal tract, periarteritis nodosa, cirrnosis of the liver, and nephrosclerosis. These diseases were equally common in the experimental and control animals. Orig. art. has 6 tables.

ASSOCIATION: none

SUBMITTED: 10Apr64 ENCL: 00 SUB CODE: LS

NO REF SOV: 000 OTHER: 000

L 23040-65 EPF(e)/EWP(f)/EWT(m) Pc-4/Pr-4 EM ACCESSION NR: AP4044081 S/0180/64

S/0189/64/000/004/0090/0090

AUTHORS: Nifant'yev, E.Ye.; Fedorov, S. Q.

TITLE Phosphites and phosphonites of novolac resins.

SOURCE: Moscow. Universitet. Vestnik. Seriya 2. Khimiya, no. 4, 1964, 90

TOPIC TAGS: novolac phosphite, novolac phosphonite, novolac resin, phosphorylation; intermolecular transphosphorylation; polyphosphite, polyamidophosphite, polyphosphite oxidation, polyphosphate, polycoxyalkylphosphonate, aminomethylphosphonate, curing, urotropine

ABSTRACT: Novolac resins were phosphorylated with esters and amides of phosphorous and phosphonic acids. A novolac (molecular weight o30) reacted quantitatively at 150-1700 with a three-fold excess of the monoethyl ester of methylphosphonic acid to form a phosphorylated resin I. Similar reaction at 160-1700 with dimethylphosphite gave an organic solvent-soluble resin II containing segments of arylmethylphosphite. On heating above 1700 II underwent intermolecular transphosphorylation to form an insoluble vitreous polymer, which may also be formed by conducting the phosphorylation reaction at 180-1903.

L 23040-65 ACCESSION NR: AP4044081

Phosphorylation of the novolac at 140-150C with di-\$\beta\$-chlorethyl-phosphite gave polyphosphites which may, or may not, be cross-linked. The hexaethyltriamide of phosphorous acid reacted with novolac at 120-140C to form a soluble polyamidophosphite which was transformed to a three-dimensional polyphosphite on prolonged standing in vacuum. The polyphosphites and polyphosphonites formed were very reactive. II was oxidized to the polyphosphate. I and II reacted with aldehydes and tetraethyldiaminomethylene to form poly-\$\alpha\$-oxyalkyl- and \$\alpha\$-aminomethylphosphonates. Liquid I and II resins were cured with urotropine to infusible resins. Orig, art. has: no graphics.

ASSOCIATION: Kafedra khimicheskoy tekimologii Moskovskego gosudarstvennogo universitet (Department of Chemical Technology, Moscow State University)

SUBMITTED: 10Apro4

ENCL: 00

SUB CODE: MT, OC

NR REF SOY: 000

OTHER: 000

Cord 2/2

PETHCV, K.A., NIPAMTIVEV, F.Ne., RKINTRUA, R.M., EGRRETY, J.M.

Anaphorise-containing segments for large segments analogs of prosphorus-containing loss of segments.

Vysokom.soed.r.no.5: Mi-933 My 'Oa. Mi9A 17:0

L 10683-85 ENT(m)/EFF(c)/ENP(j) Pc-4/Pr-4 ER

ACCESSION NR: AP4045417 S/0190/64/006/009/1545/1549

AUTHOR: Petrov, K. A.; Nifant'yev, E. Ye.; Gol'tsova, R. G.

TITLE: Synthesis of polyphosphita-based neutral polyphosphates and amidophosphates

SOURCE: 'Vy*sokomolelulysrny*ye soyedineniya, v. 8, no. 9, 1964, 1845-1849

TOPIC TAGS: polyphosphate, neutral polyphosphate, amidophosphate, polyphosphite, phosphorylated polymer

ABSTRACT: Neutral polygleylenephosphates (mol. wt. 12000 - 15000 determined from light dispersion in dimethylformamide) were prepared either by the chlorination of polyallylenephosphites (mol. wt. 3000) to polyalkylenechlorophosphates with subsequent esterification of the products with elcohols, by the reaction of polyphosphites with p-quinone:

[-0-F(0H)-0-(CH₀)-1/2 - 0-(CH₀)-1/2 -

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r by the rea	tion of polyalkylenephospi and a fertiary amine:	ites with alcohol	ls in the presenc	e of carbon	
	[—0—P(OH)—0—(СНа)а	RON+COL.) = P = 0 = (CHA) = OR		,
midophosph hylenechloro phosphate, th nexamethylen ester of poly nexamethylen he hexameth	g primary or secondary a les were also prepared. Shosphate, poly-3-oxaamy a p-hydroxy phenyl ester estutylphosphate, neutral pexamethylenephosphoric acid, the dibulenediamine-based amide of polyhexamethyleneoxyphenylamide oxyphenylamide oxyphenyl	The 12 products vienechlorophosp of polyhexamethyler acid, the beta-dictylamide of polyhexamethyler phosphoric acid, amethylenephosp	discussed are: hate, polyhexam vienephosphoric nephosphate, the ethylaminoethyl hexamethyleneph nylenephosphoric and the diethyla	ethyleneethyl- acid, poly- beta-cyncethyl ester of poly- nosphoric acid, acid, the delta- ammonium salt procedures and	1

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Card \$/8							

PETROV, K.A.; NIFANT'YEV, E.Ye.; KHORKHOYANU, L.V.; SHCHERBA, I.G.

Phosphites and phosphinites of triols and their derivatives. Zhur.ob. khim. 34 no.1:70-77 Ja '64. (MIRA 17:3)

FETROV, K.A.; NIFANT'YEV, E.Ye.; SHCHEGOLEV, A.A.; TUSEYEV, A.P.

1,2,3,4-Diisapropylidenegalactose 6-methyl phosphinite. Zhur.ob.khim.
34 no.21690-693 F '64. (MIRA 17:3)